

Chemistry 1411 Chapter 1

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General Chemistry I (CHEM 1411) Chemistry-Study of change: Atomic structure of elements, chemical bonding and chemical reactions, molecular structure, electronic configurations and periodic relationships among the elements, thermodynamics, states of matter, and properties of solutions.

Chemistry 1411 Exam 1

CHEM 1411 - General Chemistry I Practice Problems, Chapters 1-3 Chapter 1 - Chemistry: The Study of Change 1. Element, compound, homogeneous mixture (solution), or heterogeneous mixture: a) orange juice b) brass c) 0.9% saline (NaCl) solution (freshly-squeezed) d) garden soil e) room air f) methane gas g) sodium metal h) N₂ gas i) Cu(NO₃)₂ crystals 2

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CHEM 1411 SAMPLE EXAM # 1 - KEY (1, 2, and 3) PART-I 1. B (pico = 10⁻¹², 134 x 10⁻¹² m = 1.34 x 10⁻¹⁰ m) 2. C C = 5/9(F-32) = 5/9(244.2 - 32) = 5/9(212.2) = 117.9 oC 3. A 4. D sum = 26.2942 g , round off to the least number of decimal point (one), sum = 29.3 g 5. D 6. A 7. B LiNO₃ is Lithium nitrate 8.

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Chemistry 1411 Practice Exam 1, Chapters 1-4 Chang

Chemistry; Chemistry 1411; Hefley; Chapter 1; Nikki C. • 31 cards. IUPAC; International Unit of Pure and Applied Chemistry - decide name of new element based on scientist/country; Groups. 18 vertical columns on periodic table, show elements with similar chemical properties. Period. 7 horizontal rows on periodic table ...

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